## SAMPLE PAPER I

## $\underline{X}$ - SCIENCE (THEORY)

Time: $\mathbf{2}^{11 / 2}$ Hours
Max. Marks : 60

## General Instructions

1. The question paper comprises of two sections A and B. You have to attempt both the sections.
2. All questions are compulsory.
3. There is no overall choice. However, internal choice has been provided in all the three questions of five marks category. Only one option in such questions is to beattempted.
4. All questions of section $A$ and all questions of section $B$ are to be attempted separately.
5. Questions 1 to 6 in section $A$ and 17 to 19 in section $B p$ are very short answer questions.These carry one mark each.
6. Questions 7 to 10 in section A and 20 to 24 in section B are short answer type questions and carry two marks each.
7. Questions 11 to 14 in section A and 25 to 26 in section B are also short answer type questions and carry three marks each.
8. Questions 15 and 16 in section A and question 27 in section B are long answer type questions and carry five marks each.

## SECTION A

1. A ray of light LM is incident on a mirror as shown in the figure. The angle of incidence for this ray is the angle between it and the line joining two other points in the figure. Name these two points.

2. Metals generally occur in solid state. Name and write symbol of a metal that exists in liquid state at room temperature.
3. one reason.
4. The following table gives the values of refractive indices of a few media.

| S.No. | 1 | 2 | 3 | 4 | 5 |
| :--- | :---: | :---: | :---: | :---: | :---: |
| Medium | Water | Crown Glass | Rock salt | Ruby | Diamond |
| Refractive <br> Index | 1.33 | 1.52 | 1.54 | 1.71 | 2.42 |

Use this table to give an example of a medium pair so that light speeds up when it goes from one of these media to another.
5. Why do potato chips manufacturers fill the packet of chips with nitrogen gas?
6. Generally alloys are used in electrical heating devices instead of pure metals. What could be the reason?
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7. Tooth enamel is one of the hardest substances in our body. How does it undergo damage due to eating chocolates and sweets? How do tooth pastes prevent this damage?
8. A student has been collecting silver coins and copper coins。One day she observed a black coating on silver coins and green coating on copper coins. Give the chemical name of black and green coating. How are they formed?
9. Two students perform the experiments on series and parallel combinations of two given resistors $R_{1}$ and $R_{2}$ and plot the following $V-I$ graphs.

B


Which of two diagrams correctly represent the labels 'series and parallel' on the plotted curves? Justify your answer.
10. Draw the pattern of magnetic field lines of a current carrying solenoid. What does the pattern of field lines inside the solenoid indicate? Write one application of magnetic field of current carrying solenoid.
11. Out of two solar cookers, one was covered with a plane glass slab and the other was left open. Which of the two solar cookers will be more efficient and why?
12. What is biogas? How is it obtained from biomass? Why is biogas considered an ideal fuel?
13. A student finds the writing on the blackboard blurred and unclear when sitting on the last desk in the classroom. He however, sees it clearly when sitting on the front desk at an approximate distance of 2 m from the blackboard.
Draw ray diagrams to illustrate the formation of image of the blackboard writing by his eyelens when he is seated at the (i) last desk (ii) front desk.
Name the kind of lens that would help him to see clearly even when he is seated at the last desk. Draw a ray diagram to illustrate how this lens helps him to see clearly.
14. What is meant by reactivity series of metals? State which of the following chemical reactions will take place giving suitable reason for each.
$\mathrm{Zn}(\mathrm{s})+\mathrm{Cu} \mathrm{SO}_{4}(\mathrm{aq})---->\mathrm{Zn} \mathrm{SO}_{4}(\mathrm{aq})+\mathrm{Cu}(\mathrm{s})$
$\mathrm{Fe}(\mathrm{s})+\mathrm{Zn} \mathrm{SO}_{4}(\mathrm{aq}) \quad--->\mathrm{Fe} \mathrm{SO}_{4}(\mathrm{aq})+\mathrm{Zn}(\mathrm{s})$
$\mathrm{Zn}(\mathrm{s})+\mathrm{Fe} \mathrm{SO}_{4}(\mathrm{aq})---->\mathrm{Zn} \mathrm{SO}_{4}(\mathrm{aq})+\mathrm{Fe}(\mathrm{s})$
15. (a) Two lenses have power of (i) +2 D (ii) -4 D . What is the nature and focal length of each lens.?
(b) An object is kept at a distance of 100 cm from each of the above lenses.

Calculate the (i) image distance (ii) magnification in each of the two cases.
16. The following table shows the position of six elements $\mathrm{A}, \mathrm{B}, \mathrm{C}, \mathrm{D}, \mathrm{E}$ and F in the periodic table.

| Groups <br> Periods | 1 | 2 | 3 to 12 | 14 | 14 | 15 | 16 | 17 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 2. | A |  |  | 18 |  |  |  |  |
| 3. |  | D |  |  |  | B |  |  |

Using the above table answer the following questions :
(a) Which element will form only covalent compunds?
(b) Which element is a metal with valency 2?
(c) Which element is a non-metal with valency 3?
(d) Write a common name for the family of elements C and F .
(e) Out of D and E , which one has a bigger atomic radius and why?
17. An organic compound ' A ' is widely used as a preservative in pickles and has a molecular formula $\mathrm{C}_{2} \mathrm{H}_{2} \mathrm{O}_{2}$. This compound reacts with ethanol to form a sweet smelling compound ' B '.
(i) Identify the compound ' A '
(ii) Write the chemical equation for its reaction with ethanol to form compound ' B '.
(iii) How can we get compound ' A ' back from ' B '?
(iv) Name the process and write corresponding chemical equation.
(v) Which gas is produced when compound 'A' reacts with washing soda?

Write the chemical equation.
OR
(a) Why does carbon form largest number of compounds? Give two reasons.
(b) Why are some of these called saturated and other unsaturated compounds?
(c) Which of these two is more reactive?
(d) Write the names of the compounds :

(i) $\mathrm{CH}_{3}-\mathrm{CH}_{2}-\mathrm{Br}$
(ii) $\mathrm{H}-\mathrm{C}-\mathrm{C}-\mathrm{C}-\mathrm{C}-\mathrm{C} \equiv \mathrm{C}-\mathrm{H}$
18. In a household electric circuit different appliances are connected in parallel to one another. Give two reasons.

An electrician puts a fuse of rating 5A in that part of domestic electrical circuit in which an electrical heater of rating $1.5 \mathrm{~kW}, 220 \mathrm{~V}$ is operating. What is likely to happen in this case and why? What change, if any, needs to be made?

You are given following current-time graphs from two different sources:


(i) Name the type of current in two cases.
(ii) Identify any one source for each type of these currents.
(iii) What is the frequency of current in case II in India?
(iv) Use above graphs to write two difference between the current in two cases.
(v) What is the advantage of AC over DC?

## SECTION B

19. Why do aquatic animals breath faster that the terrestrial animals?
20. Name one plant growth hormone which retards growth during extremely dry season.
21. When plants are eaten by primary consumers, a great deal of energy is lost as heat to the environment and some amount goes in carrying out various life processes. State the average percentage of energy lost in this manner.
22. List the sequence of events that occur when a plant is exposed to unidirectionallight,deading to bending of a growing shoot. Also name the hormone and the type of movement
23. From the set of figures given below, make a pair of homologous and analogous organs each and give one reason in case of both, to justify your answer.
24. How is transport of water in xylem tissue different from translocation of food in phloem tissue?
25. A group of grass hoppers - some green and some brown lived in a grassland having dry bushes Humanfogeliss. Bat Wing Bird Wing
(a) Which one would normally be picked up by predatory birds and why?
(b) Population of which grasshopper will increase?
(c) Name this phenomenon
Q. 26 Make an aquatic food chain up to tertiary consumer level. State the trophic level at which concentration of pesticide is maximum and why?
Q. 27 Identify male and female in the figures given below. Also fill in the blanks 3 to 8 and then clarify about the misconception that mother and not father is responsible for bearing daughters and not sons.

